Inorganic Chemistry

ADDITION/CORRECTION

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Correction to Monofluoride Bridged, Binuclear Metallacycles of First Row Transition Metals Supported by Third Generation Bis(1-pyrazolyl)methane Ligands: Unusual Magnetic Properties

[*Inorganic Chemistry* **2009**, *48*, 10658–10669 DOI: 10.1021/ic901352p].

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Page 10659. The assumption made in this paper that the gelatin capsules used to hold the samples in the SQUID measurements "make a negligible contribution to the overall magnetism" is incorrect for the $3d^9$, strongly antiferromagnetically coupled compound $[Cu_2(\mu\text{-}F)(\mu\text{-}m\text{-}[CH(pz)_2]_2C_6H_4)_2](BF_4)_3$ (**4**). Analysis of a new data set[1] in which the susceptibility of the sample container was measured independently and the data were corrected appropriately yields the results shown in corrected Figure 8 (p 10666) where, using the Hamiltonian $\mathcal{H} = -2JS_1 \cdot S_2$, the J value (Abstract on p 10658 and pp 10665 and 10668) is calculated as $-181.3(4)$ cm^{-1} with a g value[2] of 2.153(3). The Cu(II) ions in compound **4** are strongly antiferromagnetically coupled, but not to the degree indicated in the original paper. Table 4 (p 10668) is also corrected as shown below.

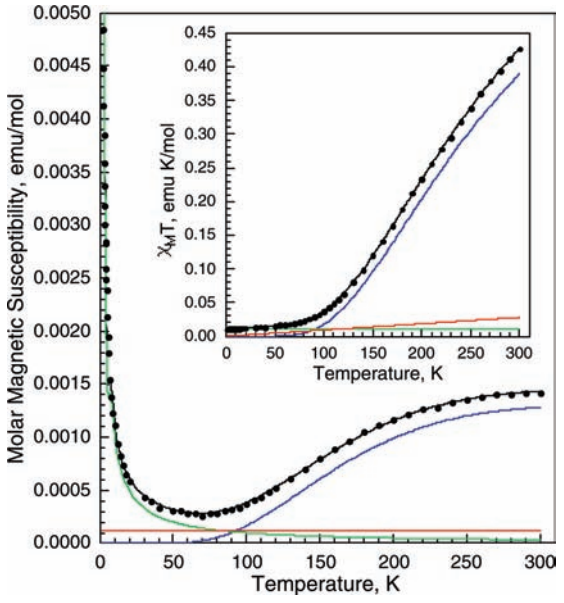

Figure 8. Molar magnetic susceptibility of $4 \cdot 2H_2O$ versus temperature. Inset: $\chi_M T$ versus temperature. Black: data points and total fit. Blue: dimer fit with $S_1 = S_2 = {}^1/_2$, $g = 2.153(3)$, and $J = -181.3(4)$ cm^{-1}. Red: second-order Zeeman contribution, $N\alpha = 0.000\,093(6)$ emu/mol. Green: 2.72(4)% of an unknown paramagnetic impurity that is assumed to have $g = 2.00$ and $S = {}^1/_2$.

Table 4. Solid State Magnetic Properties

complex	J, cm^{-1}	g	D, cm^{-1}	E, cm^{-1}
$[Fe_2(\mu\text{-}F)(\mu\text{-}L_m)_2](BF_4)_3$, **1**	$-10.4(2)$	2.12(2)−2.15(2)	$-10.2(3)$	$-2.0(5)$
$[Co_2(\mu\text{-}F)(\mu\text{-}L_m)_2](BF_4)_3$, **2**	$-0.67(5)$	2.45(1)	$\pm 61(2)$	
$[Cu_2(\mu\text{-}F)(\mu\text{-}L_m)_2](BF_4)_3$, **4**	$-181.3(4)$	2.153(3)		

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REFERENCES

(1) The magnetic susceptibility has been remeasured using a MPMS-XL-5 SQUID magnetometer at the Faculty of Chemistry, Wroclaw University, Poland.

(2) The g value was verified by EPR analysis: Ozarowski, A., unpublished results.

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